4531

Comment on "Spin-Coated Periodic Mesoporous Organosilica Thin Films with Molecular-Scale Order within the Organosilica Wall"

The recent publication in this journal of work by Wu et al.¹ reports preparation of mesoporous benzene-silica thin films with molecular-scale periodicity. However, we think the experimental results do not properly support the conclusion that the pore walls consist of the periodically arranged benzene-bridged bissilanetriols (Ph-triols) but indicate that the obtained film is a blend of a mesoporous thin film with amorphous pore walls along with a small fraction of nonporous Ph-triol particles having molecular-scale periodicity. First of all, few T⁰ (R-Si(OH)₃: triol) species are detected in the ²⁹Si MAS NMR spectrum (Figure 3), indicating the triol groups are in minor extent, contradicting the authors' conclusion. Additionally, such a low degree of condensation (mainly T² species) is typical for organosilica materials prepared by sol–gel synthesis under acidic conditions² and

not a favorable reason to elongate *d*-spacing of the molecularscale periodicity to 10.1 Å. Second, the XRD peaks at 10° $< 2\theta < 50^{\circ}$ due to periodic arrangement of the Ph-triols (Figure 1b,c) are very sharp despite the fact that the molecular-scale periodicity in the thin pore walls (e.g., a few nanometers) usually results in broadening of the XRD peaks,³ suggesting that the periodic arrangement of the Ph-triols is not formed in the pore walls. Finally, the lattice fringes appear to run independently of the pore channel patterns and also change direction in the same channels⁴ in the TEM image (Figure 4b), which is not likely to occur in mesoporous materials with molecular-scale periodicity and may result from overlapping of two phases.

Takao Tani and Shinji Inagaki*

Toyota Central R&D Laboratories, Inc., Nagakute, Aichi 480-1192, Japan

CM7035844

^{*} Tel: +81-561-71-7393. Fax: +81-561-63-6507. E-mail: inagaki@ mosk.tytlabs.co.jp.

⁽¹⁾ Wu, S.-Y.; Hsueh, H.-S.; Huang, M.-H. Chem. Mater. 2007, 19, 5986.

 ^{(2) (}a) Shea, K. J.; Loy, D. A.; Webster, O. J. Am. Chem. Soc. 1992, 114, 6700. (b) Fan, H. Y.; Lu, Y. F.; Stump, A.; Reed, S. T.; Baer, T.; Schunk, R.; Perez-Luna, V.; Lopez, G. P.; Brinker, C. J. Nature 2000, 405, 56.

⁽³⁾ Inagaki, S.; Guan, S.; Ohsuna, T.; Terasaki, O. Nature 2002, 416, 304.

⁽⁴⁾ The lattice fringes run from top-left to bottom-right in the upper part whereas they run from bottom-left to top-right in the lower parts in the same pore channels (see the left side of Figure 4b).